O Level Pure Chemistry MCQs

Organic Chemistry Test 1.0

Q1

Two hydrocarbons have the same percentage by mass of carbon and hydrogen.

Which statement is true for both hydrocarbons?

- A They belong to the same homologous series.
- B They have similar chemical properties.
- **C** They have the same empirical formula.
- D They have the same molecular formula.

Q2

In which reaction is water not a product?

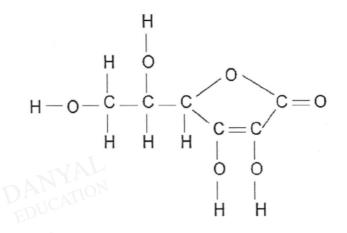
- A combustion
- B esterification
- **C** fermentation
- D neutralization

Q3

In the polymerization of ethene to polyethene, there is no change in

- A boiling point.
- B density.
- C empirical formula.
- D molecular formula.

The structural formula of an organic compound is shown below.





Which substance(s) can react with this compound?

- I acidified potassium manganate(VII)
- II aqueous bromine
- III sodium carbonate
- A I and II only
- B I and III only
- C II and III only
- D I, II and III

Q5

A student investigated the reaction of different vegetable oils with hydrogen.

100 cm³ of hydrogen was passed through 1 g samples of vegetable oils containing a suitable catalyst. The volume of hydrogen gas remaining in each reaction was recorded in the table below.

	vegetable oil	volume of hydrogen remaining (cm ³)
EDI	P .	100
	Q	87
	R	0



Which vegetable oil(s) is/are unsaturated?

- A P only
- B P and Q
- C Q and R
- D R only

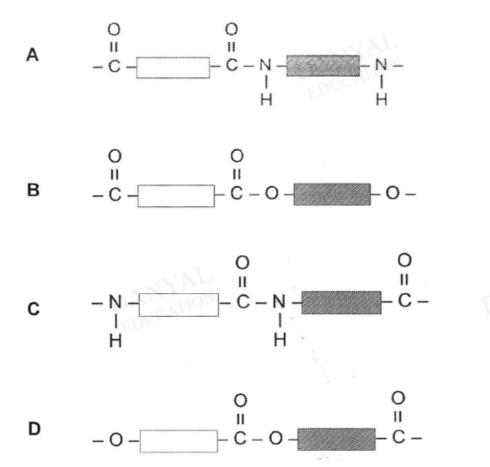
The reaction between a carboxylic acid, C_xH_yCOOH and an alcohol, $C_zH_{2z+1}OH$ produces an ester.

How many carbon, hydrogen and oxygen atoms are there in this ester?

	number of carbon atoms	number of hydrogen atoms	number of oxygen atoms
Α	x + z	y + 2z	2
в	x + z	y + 2z + 1	DAN 2N
С	x + z + 1	y + 2z	EDUCA 2
D	x + z + 1	y + 2z + 1	2

Q7

What is the correct structure of nylon?



The reaction between the hydrocarbon C_xH₆ and hydrogen can be represented by the equation:

 $C_xH_6(q) + H_2(q) \rightarrow C_xH_8(q)$

Which of the following statements about the above reaction is true?

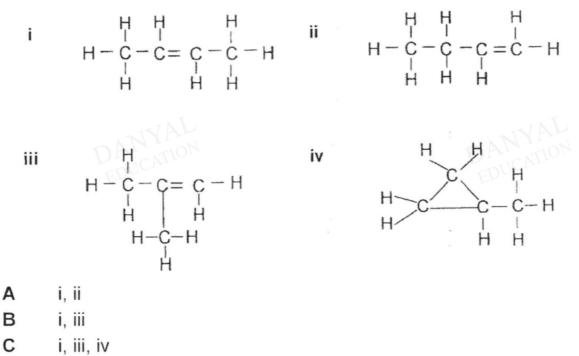
- It is a substitution reaction. Α
- UV light is required for the reaction to take place. В
- The molecular formula of the hydrocarbon is C_2H_6 . С
- The molecular formula of the hydrocarbon is C₃H₆. D

09

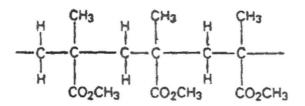
The diagram shows the structure of a hydrocarbon X.

$$\begin{array}{cccccc} H & H & H & H \\ I & I & I & I \\ H - C = C - C - C - H \\ I & I \\ H & H \end{array}$$

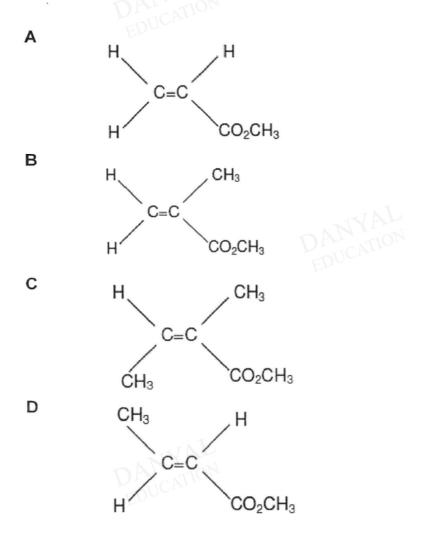
Which of the following structures are isomers of hydrocarbon X?



The structure of the plastic Perspex is shown below.



What is the molecular structure of the monomer from which this plastic is formed?



Answers

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Q1 C Q2 C Q3 C Q4 A Q5 C Q6 D Q7 A Q8 D

- Q9 C
- Q10 B