O Level Combined Chemistry MCQs

Organic Chemistry - Alkanes Test 1.0

Q1

The general formula of the alkanes homologous series is C_nH_{2n+2}.

Which physical property decreases as n increases?

- A boiling point
- B flammability
- C melting point
- D viscosity

Q2

Which statement about alkanes is correct?

- A Ethane reacts with chlorine in an addition reaction.
- B Propane has a higher boiling point than butane.
- C The molecule of the alkane that contains 99 carbon atoms has 200 hydrogen atoms.
- D Ethane is the first member of the alkanes series.

Q3

Which equation represents a cracking process?

- A C₄H₈ + H₂→ C₄H₁₀
- B C₈H₁₈ → 2Ċ₃H₆ + C₂H₄ + H₂
- C C₆H₁₂O₆→ 2C₂H₅OH + 2CO₂
- D C₄H₉OH → C₄H₈ + H₂O

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Q4

The reaction between a hydrocarbon, C_nH₆, and chlorine can be represented by the equation:

$$C_nH_6(g) + CI_2(g) \rightarrow C_nH_5CI(g) + HCI(g)$$

Which statement about the reaction is correct?

- A It is an addition reaction.
- B Nickel is used as a catalyst in the reaction.
- C The molecular formula of the hydrocarbon is C₃H₆.
- D UV light is required for the reaction to take place.

Q5

What are all the possible combustion products of methane?

- A carbon, carbon dioxide, carbon monoxide and water
- B carbon, carbon monoxide and hydrogen
- C carbon dioxide, carbon monoxide and hydrogen
- D carbon dioxide, carbon monoxide, hydrogen and water

Q6

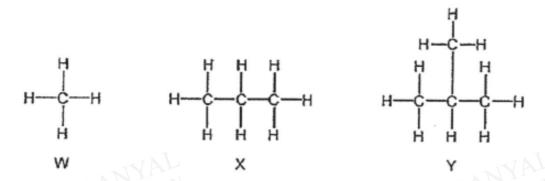
Which of the products, $C_{12}H_{24}$ and H_2 , could be formed by cracking dodecane, $C_{12}H_{26}$?

	C12H24	H ₂
Α	x	х
В	DX	1
C	7	x
D	4	√



Q7

The structure of three hydrocarbons from the same homologous series are shown.



Which statement is correct?

- A W has the lowest boiling point.
- B X and Y have the same molecular formula.
- C All three molecules are unsaturated hydrocarbons.
- D All three molecules have different chemical properties.

Q8

Butane undergoes the first substitution reaction with chlorine in the presence of ultraviolet light. A student suggested that the following organic products are formed.

- 1 CH₃CHC/CHC/CH₃
- 2 CH₃CH₂CH₂CH₂CI
- 3 CH₃CHC/CH₂CH₃
- 4 CCI₃CCI₂CCI₂CCI₃

Which of these four products are formed during the first substitution reaction?

A 1 and 2

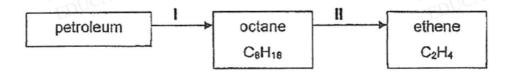
C 1, 2 and 3

B 2 and 3

D 1, 2, 3 and 4

Q9

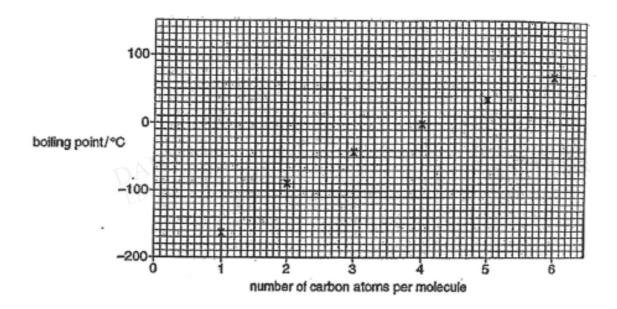
The diagram shows two stages in a reaction scheme.



What is the name of stage II?

- A Substitution
- B Fractional Distillation
- C Hydrogenation
- D Cracking

The graph shows how the boiling point of some hydrocarbon depends on the number of carbon atoms in their molecules.



Which hydrocarbon is an alkane with a boiling point close to 0 °C?

Answers

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Q1B

Q2 C

Q3 B

Q4 D

Q5 A

Q6 D

Q7 A

Q8 B

Q9 D

Q10 B

DANYAL

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